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Aatmgyan Ki Vidhiyan Book,

श्री श्री रवि शंकर (Sri Sri

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शंकर (Sri Sri Ravi Shankar)

Q: Why doesn't the water boil
in my kettle but the stovetop
does? When I boil water in a

kettle it takes a long time to get to a boil. When I boil water in a pan on the stovetop it boils in the same time. Why is this happening? A: You get the right amount of heat when the water is in contact with the surface of the pan, which is the case in the stovetop, but not when it's in contact with the entire volume. This happens when you boil a pot

in the stove, as it's a "closed" system. The water is contained in a chamber which heats it, and as long as there's heat energy in the chamber, the water will continue to heat and boil. When you boil a pot in a kettle, there is no chamber to contain the heat, and so the water has to heat the entire volume it's in contact with, which takes

much longer. A: I'll give you the more complex explanation (and my personal opinion).

When you boil water in a pot, the heat gets to the water through the walls of the pot.

(at least with a pot made from a metallic material, where the heat is transported by

conduction, but it is the same principle). When you boil

water in a kettle, the heat gets

to the water through the walls of the kettle (and the lid).

This is the reason why a pot is "hotter" than a kettle. You can put your finger on a pot and you will feel the heat through the wall of the pot. If you touch the inside of a kettle, you will notice that it's warmer than you expected. To make it simpler, the kettle is like an oven and
